-2nd half (1)-12-14

Phulmary / Sem-I/ATKT/ Binslit Chemistry

OldCoule

CN-2 Con. 9504-12. (2 Hours)

CN-2876

[Total Marks: 40

3

4

2

N.B.:(1) Question No. 1 is compulsory.

- (2) Attempt any four questions from remaining five questions.
- 1. (a) Give IUPAC nomenclature for the following: 3
 - (i) [(Co (NH₃)₅ Cl] Cl₂
 - (ii) $[Cr(NH_3)_6]^{+3}$
 - (iii) K [Pt Cl₃ (NH₃)]
 - (b) Write the structures of the following:-2
 - (i) Tetra ammine aquamonobromo-cobalt (III) nitrate
 - (ii) ferric hexa cyano ferrate (II)
 - (c) Define the following:-
 - (i) Transition state
 - (ii) Astringent
 - (iii) Effective Atomic Number [EAN].
- 2. (a) Discuss specific acid catalysis. 4
 - (b) Elaborate the concept of reactivity and selectivity for chemical reactions of halogenation of alkanes.
- 3. (a) Classify types of antacids with example. Elaborate on combination antacids. 4
 - (b) Explain the effect of temperature in rate of reaction and what is its significance. 4
- 4. (a) Write a note on initial rate kinetics. 4
 - (b) Discuss with example effect of weak field and strong field ligand on crystal field splitting in case of co-ordinate complexes.
- 5. (a) Give property and use of following (any two):-4
 - (i) Hydrogen peroxide
 - (ii) Talc
 - (iii) Povidone iodine
 - (b) Discuss the effect of primary isotope.
- 6. (a) Write short notes on (any two) :-
 - (i) Charge transfer complex
 - (ii) Microscopic reversibility
 - (iii) Mechanism of antimicrobial agents
 - (b) Predict the magnetic property and geometry of [Fe (H₂O)₆]⁺³