Ph | Sem-I | ATKT | O. C.-I - Old (2 Hours) 12 | 12/12 | Total Marks: 40

Con. 9510-12.

N.B.: (1) Question No. 1 is compulsory.

- (2) Answer any four questions out of the remaining six questions.
- (3) Give figures and examples wherever necessary.
- 1. (a) Give IUPAC nomenclature for the following organic compounds.

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- (ii) (CH<sub>3</sub>)<sub>3</sub> C CH<sub>2</sub> C(CH<sub>3</sub>)<sub>3</sub>
- (b) Give suitable structures for the following:-

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- (i) 4 (1–methyl ethyl) heptane
- (ii) Neopentyl bromide.
- (c) Draw resonating structures for the following:-

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(i) 
$$c = 0$$
(ii)  $OCH_3$ 

- 2. (a) Compare the acidity or basicity between following pairs of compounds and justify:
  - (i) Phenol, 2, 4 dinitrophenol
  - (ii) Aniline, Triethyl amine.
  - (b) Indicate whether the following compounds are nucleophiles or electrophiles:-

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(i) 
$$(CH_3)_2 NH$$

(i) 
$$\left(CH_3\right)_2 NH$$
 (ii)  $\left(CH_3\right)_2 \overset{\oplus}{C}H$ 

- (c) Combustion of a 5·17 mg sample of a compound gives 10·32 mg of CO<sub>2</sub> and 4·23 mg 3 of H<sub>2</sub>O. The mol wt is 88. What is the empirical formula of the compound?
- 3. Write short notes (any four) :-

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- (a) Hyperconjugation
- (b) Role of solvent in  $S_N^{-1}$  reaction
- (c) Dimerization of alkenes
- (d) Inductive effect
- (e) Saytzeff Orientation.

4. (a) Write the structures of the products formed and complete the following reactions:-



- (b) Give one example each for the following reactions:-
  - (i) Halohydrin formation
  - (ii) Dehydrohalogenation of alkyl halides
  - (iii) Hydration of alkynes
  - (iv) Oxymercuration Demercuration.
- 5. (a) Discuss the mechanism and stereochemistry of  $S_N^2$  reaction with a suitable example.

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- (b) Discuss methods of preparation of alkyl halides from alkenes with examples.
- 6. (a) Explain the mechanism of Unimolecular elimination of alkyl halides and give suitable evidence for the same.
  - (b) Carry out the following conversion: Propanol to Allyl bromide.
  - (c) Give probable structure for an alkene yielding propionaldehyde and acetone on hydrolysis.
- 7. (a) On the basis of hybridisation, explain the formation of carbon-carbon single bond.
  - (b) Discuss the mechanism of addition of halogens to alkenes.
  - (c) Give the structures of A and B in the following reaction sequence.

Butan-2-01 
$$\xrightarrow{\text{H }_2\text{SO}_4}$$
  $\stackrel{\text{(A)}}{\text{(A)}}$   $\xrightarrow{\text{B }_2/\text{C C }_1}$   $\stackrel{\text{(B)}}{\text{(B)}}$