58-mk: 1stHf-11.

Con. 2654-11.

19/5/11(R)

## Phaam. Anal. IV (REVISED COURSE)

(2 Hours)

Jen Mile 10 Total Marks: 40

Question No. 1 is compulsory. Attempt any four questions from the remaining six questions. Figures to the right indicate full marks. Draw neat labelled diagrams wherever necessary. (a) Explain the following terms: (any three) Gradient elution Confidence Limit (ii) Retention time Name the following: (any two) Two solvents used in normal phase HPLC. Two visulisation techniques used for TLC. A gas chromatography detector used for sample containing halides. Two ion-pairing agents used in HPLC. Classify chromatographic techniques schematically. Justify the advantage of using gas chromatography as an analytical technique. Enlist the various parameters for analytical method validation as per U.S.P. Explain any two parameters in detail. 3. (a) Draw a block diagram depicting the instrumentation for HPLC. Explain any one detector used in HPLC in detail. Discuss any four factors that affect resolution in TLC. 4. (a) Give the objectives for sampling of raw materials. Discuss the different sampling schemes. (b) Explain the principle and application of Thermal conductivity detector used in gas chromatography. (a) Discuss the various ion-exchange resins used in chromatography. (b) Explain the principle for thermogravimetric analysis and give its pharmaceutical applications. (a) A sample of drug 'X' was analysed and the concentration (mg/ml) obtained after colorimetric analysis was as follows:--5.61 5.59 5.60 5.64 5.62 5.55 5.58 5.70 5.48 5.58 Use t-test to see if the true mean concentration of the drug can be 5.60 mg/ml at 95% confidence interval. Given: The value of t at 95% confidence interval and 9 degrees of freedom is 1.83. What is meant by Thermal analysis? Explain Differential Scanning Calorimetry (DSC) in detail. Write short notes on (any two):-HPTLC Development techniques in paper chromatography Size exclusion chromatography.