Q.P. Code: 3065

(OLD COURSE) (2 Hours)

[Total Marks: 75

N.B.	: (1) Question No. 1 is compulsory.	
	(2) Attempt any four from remaining six questions.	
	(3) Figures to the right indicate full marks.	
	(4) All questions carry equal marks.	N
	((5) Atomic weight: $H = 1$, $C = 12$, $N = 14$, $O = 16$, $Na = 23$, $Mg = 24$, S	
		= 32, $Cl = 35.5$, $Ca = 40$, $Ba = 137.5$.	
1.	Att	empt any five from the following:	15
		a) How volatile matters are determined from the coal sample? Write the significance of volatile matter analysis.	
	(1	b) Write the composition, properties and uses of duralumin	
		c) Write the classification of composite materials.	
	,	d) Write the difference between catholic and anodic coating.	
		e) Define and explain activation energy.	
		f) Write a short note on "E green propellant".	
2.	(a)	What are fibre reinforced composite materials. Write their classification.	6
۷.	(b)	1 1 0 000 77 00/ 10	5
	(-)	= 7%, S = 0.6% and remaining ash. Calculate H.C.V. and L.C.V. using Dulong's	
		formula.	
	(c)	Write a short note on "antiknock agents."	4
	(-)	C. C	
3.	(a)	Explain traditional and green route of synthesis of indigo dye.	6
		What is stress corrosion? Explain stress corrosion with the example of season	5
	(0)	cracking.	
	(c)	What are the applications of powder metallurgy.	4
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4.	(a)	What is catalysis. Explain the adsorption theory of heterogenous catalysts.	6
1.		Calculate the weight of air needed for complete combustion of 1kg of coal	5
	(0)	containing $C = 65\%$, $H = 7\%$, $O = 9\%$, $S = 1\%$ and remaining nitrogen.	
	(c)		4
	10	Cargulate the percentage atom coolienty for the fortest this reaction with respect	-7

 $C_6H_6 + CH_3Cl \underline{AlCl_3} C_6H_5CH_3 + HCl$

Benzene Methyl Toluene Chloride

TURN OVER

to product toluene.

5.	(a)	How these factors influence the rate of corrosion. (i) pH of the medium,	6
		(ii) Relative areas of anode and cathode,	
		(iii) Overvoltage.	0
	(b)	Write a short note on zeolite as a catalyst.	5
	(c)	Wate are laminar composites.	4
6.	(a)	What are ceramic powder? Write the manufacture of alumina.	6
	(b)	Write the transesterification reaction of production of bio-diesel. Explain advantages of bio-diesel.	5
	(c)	Explain in detail anodic protection method of corrosion control	4
7.	(a)	Explain concentration cell corrosion with the help of suitable example.	6
	(b)	What is cracking. Explain fixed bed catalytic cracking with the help of a neat diagram.	5
	(c)	Explain "prevention of waste" principle of green chemistry with the help of suitable examples.	4