CBSGS Sem-II SUB-POC

Q.P. Code: 00231

Please check whether you have got the right question paper. N.B: 1. All questions are compulsory. 2. Answer all sub questions together. 3. Figures to right indicate full marks. Q.1 a. Explain the terms (Any 5) i) Electronic configuration iii) Symmetry elements iv) Specific base v) Activated complex vi) Charge transfer complex b. Fill in the blanks i) In a first order reaction half-life of reactant is found to be 23 minutes. The rate constant for the reaction is ii) Lewis structure for H ₂ SO ₄ is			[Time: Three Hours]	arks:70]
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b. Fill in the blanks i) In a first order reaction half-life of reactant is found to be 23 minutes. The rate constant for the reaction is ii) Lewis structure for H ₂ SO ₄ is iii)			ii) Energy of activation	
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reaction is				5
ii) Lewis structure for H ₂ SO ₄ is			of reactant is found to be 23 minutes. The rate constant for the	
iii)is an example of phase transfer catalyst iv) Charge transfer complex example is				
iv) Charge transfer complex example is				
v) Ground state electronic configuration for boron is				
c. Match the following. Column A dyz Electrophilic catalysis Fe valence electrons (At.No.26) AlC13 Water Di symmetry Nucleophilic catalyst O.2 a) Draw resonating structures for: i) H ₂ PO ₄ ii) CO ₃ -2 b) Explain the formation of ethylene and clearly indicate HOMO and LUMO.				
Column A dyz Electrophilic catalysis Fe valence electrons (At.No.26) AIC1 ₃ Sp ³ hybridization Water Duclrophilic catalyst Sp ³ hybridization Water Sp ³ hybridization Water Sp ³ hybridization Water Sp ³ hybridization Water Sp ³ hybridization Sp ³ hybridization Water Sp ³ hybridization Sp ³ hy		ty Ground state electronic comigar	acion for botom is minimum.	
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d _{yz} Electrophilic catalysis Fe valence electrons (At.No.26) AlC1 ₃ Sp ³ hybridization Water pi symmetry Nucleophilic catalyst 3d ⁶ 4s ² Q.2 a) Draw resonating structures for: i) H ₂ PO ₄ ii) CO ₃ -2 b) Explain the formation of ethylene and clearly indicate HOMO and LUMO.				26
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Nucle ophilic catalyst		AIC1 ₃	Sp³ hybridization -	
Q.2 a) Draw resonating structures for : i) H ₂ PO ₄ ii) CO ₃ -2 b) Explain the formation of ethylene and clearly indicate HOMO and LUMO.				
i) H_2PO_4 ii) CO_3^{-2} b) Explain the formation of ethylene and clearly indicate HOMO and LUMO.		Nucleophilic catalyst	7 3d ⁶ 4s ²	
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b) Explain the formation of ethylene and clearly indicate HOMO and LUMO.	Q.2			2
		· = 0		
C) What is an isotone effect? Mention its significance				
The second secon				3
d) State different catalysis types and explain any one in detail.		d) State different catalysis types an	id explain any one in detail.	3
Q.3 a) Compare the geometries of NH ₃ and BH ₃ by using their group orbitals.	Q.3	a) Compare the geometries of NH ₂	and BH ₂ by using their group orbitals	3
b) With the help of Walsh diagram explain energies for linear and bent CH ₂ .	4.0			
c) State Eyring equation. Explain each term involved in it.				
d) For a reaction following first order kinetics calculate half-life if 30 % reaction gets completed in 1 2				

Q.4	b) What is QMOT? State any th	ree rules for OMOT.		3	
	c) Elaborate on first order kine			3	
	d) Explain Nucleophillic catalysi			2	
Q.5	a) Draw the resonating structu	re for the given molecules.		3	
	1) H ₂ C ⁺ CH ₃	2)	· -		
	b) Explain the formation of car	banion by molecular orbital theory		DEL SA SERVE I	
	c) A plot of lnK vs 1/T is a straig	ght line with a slope -2.8×10^2 . Calc	y. Culate the energy of activa	ation for 2	
	the reaction.			2011101 2	
	d) Give the classification of cha	rge transfer complexes with suita	ble examples.	3	
Q.5	a) Complete the following table	e on the basis of hybridization con	cent	3	
	Molecule	Hybridization state of the underlined atom	Bond angle	3	
	CH ₄	awacimied atom			#
	BF ₃				-
	PC1 ₅				
	b) Derive an expression for gen	eral base catalysis and show relev	ant bioatic obs		
	c) State and explain principle o	f microscopic reversibility	ant kinetic plots.	4	ř
	d) What is Hammond's postulat	e? Explain it by showing relevant	energy profile diagrams	2	•
		,,	By prome anabiams.	2	



CBSGS

Q.P. Code:00581

[Time: Three Hours] [Marks:70] Please check whether you have got the right question paper. N.B: 1. All questions are compulsory. 2. Draw neat, labelled diagrams wherever necessary. Q.1. a) Explain supercritical fluid state & give its applications. (03)What is optical activity? Give its applications. (02)When 0.44gm of a substance was dissolved in 22.2gm benzene, the freezing point of benzene (03)was lowered by 0.567° C. If $K_f = 5.12^{\circ}$ C, mol⁻¹, calculate the molecular weight of the substance. d) Define thermodynamics & give its applications & limitations. Classify thermodynamic systems. (04)State & Explain Faraday's law of electrolysis (03)Q.2. a) Explain Linde's method for liquefaction of gases. (04)Elaborate on Claude's method for liquefaction of gases. b) What is dielectric constant? Give its applications in pharmacy. (03)c) Explain Hess's law of constant heat summation. (04)a) Explain Ostwald & Walker's Dynamic method for measurement of relative lowering of vapour Q.3. (04)Explain efficiency of heat engine. An engine operating between 150°C & 25°C takes 500 J from (04)high temperature reservoir. Calculate the work done by it, assuming that there are no frictional losses. OR Give different statements of second law of thermodynamics. What is entropy? Explain its importance c) Explain equivalent conductance of a weak electrolyte at infinite dilution. (03)a) Derive the equation for deviation of real gases from ideal gas. Q.4. (04)b) Explain principle and working of Abbe's refractometer. (03)Explain a method to determine the molecular weight of a non-volatilve solute by elevation in (04)

P.T.O

Justify 'Depression in freezing point is a colligative property'.

boiling point.

Q.P. Code :00581

Q.5.	a)	Write a short note on polymorpl	hism.	(0	14)
	b)	What is Osmosis and describe m	odern Osmometer.	(0	14)
	c)	Define the following i) Heat of formation			3)
		ii) Heat of combustion			
		iii) Heat of solution			
		•	OR		
		Define Bond energy. Calculate Δ	H for reaction:	(0	3)
		$C_2H_4(g) + 3O_2(g) \rightarrow 2CO_2(g) + 2H_4(g)$	I ₂ O(g) from the following values		•
		Bond	Bond energies (KJ)		
		C-H	414		
		O=0	499		
		C=0	724		
		O=H	460		
		C=C	619		
Q.6.	a)	The Van der Waal's constant for a=5.57, b=0.064 Calculate the external pressure a ethane is 2.5 m ³ /Kmol and R=8.3	and internal pressure for ethane		3)
	b)	Write a short note on fractional	distillation	(0.	3)
	c)	Write a short note on Gibb's Free	e Energy	(0)	3)
	d)	State the postulates of Arrhenion	us theory of electrolytic dissocia	-	

F.Y. Bharm Sem-T(cBSGS) Sub-APP-T OP Codo: Q.P. Code:06724

1. All questions are compulsory. 2. Answer all sub questions together.

N.B:

ii)

iii)

[Time: 3 Hours]

Please check whether you have got the right question paper.

[Marks: 70]

08

3. Draw neat labeled diagram wherever necessary. Q.1 a) Answer the following. 12 i) Give functions of adipose tissue & connective tissue. ii) Define: Osmosis & passive transport iii) Give functions of Lymphatic system Explain blood transfusion reaction. Enlist any four inflammatory mediators. v) Comment on all or none principle of muscle contraction. vi) b) Define the following terms: 03 Hemostasis j) ii) Motor unit iii) Chemotaxis Q.2 a) Answer any TWO of the following: 08 $\label{thm:explain} \mbox{Explain synthesis of hemoglobin with its structure}.$ ii) Write classification of immunity. Explain in brief adaptive immunity. iii) Enlist types of leukocyte. Explain in brief arganulocytosis. b) Answer any ONE of the following: 03 Write a note on Leucopoiesis Write the pathophysiology of Myasthenia gravis. Q.3 a) Answer any TWO of the following. 08 Explain contraction of skeletal muscle in brief. i) ii) Draw a neat diagram of sarcomere. Explain the energy metabolism in the muscle. b) Answer any ONE of the following: 03 a) Explain the process of pinocytosis & phagocytosis. b) Write a note on cardiac muscle. Q.4 a) Answer any TWO of the following:

Explain the histology & functions of lymph.

Write a note on AIDS.

Write the classification of organs of lymphatic system. Explain the process of formation of

Q.P. Code:06724

	b)	Answer an	y ONE of the following:	03
		i)	Differentiate between simple & compound exocrine glands.	03
		ii)	Write a note on hypersensitivity reactions.	
Q.5	a)	Answer an	y TWO of the following:	08
		i)	Draw a diagram of plasma membrane & write its functions.	Uo
		ii)	Draw a diagram of neuromuscular junction & explain the role of acetyl choline in muscle contraction.	
		iii)	Write a note on acute inflammation.	
	b)	Answer any	y ONE of the following:	03
		i)	Explain the process of tissue repair.	03
		ii)	Write a note on active transport process.	
Q.6	a)	Answer any	TWO of the following:	08
		i)	Explain the steps involved in hemostasis & write the role of platelets in coagulation of blood.	08
		ii)	Explain the life cycle of RBC.	
		iii)	Define blood & write its composition & functions.	
	b)	Answer any	ONE of the following:	03
		i)	Explain the process of cell mediated immunity.	US
		ii)	Write a note on role of prostaglandins in inflammation.	

F.Y. B. Plann, Sem-I (CBSGS) 26/04/17 Q.P. Code: 02739

[Time: 3 Hours]

[Marks:70

Please check whether y	ou have got the right	question paper.
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1. All questions are compulsory N.B:

- 2. Draw neat diagrams wherever necessary
- 3. Figures to right indicate full marks

0.16	What are abiotic components of an ecosystem?	
Q. 1 (c	b) What do you understand by Biological Oxygen Demand.	1
()	What is "Carrying capacity"?	1
		1
	l) What is the need for studying environmental issues.	2
	e) Outline Global Environmental Crisis related to sanitation.	2
(1	What do you mean by non-point sources of water? Give examples.	2
) What are greenhouse gases? Give examples.	2
(i)	Give diagrammatic representation of carbon cycle.	2
(1)	Enlist the functional components of ecosystem.	2
Q. 2 (a	Explain food chains and food webs? Give examples.	4
(b) With a neat labelled diagram, explain Flash Steam Distillation.	4
	OR	***
(b) With a neat labelled diagram, explain the working of Photovoltaic Cell.	4
(c) What are the major causes and consequences of deforestation?	3
Q. 3 (a) Classify solid waste. Explain sanitary landfills as one of the methods of solid was management.	ste - 4
(b) Enumerate on any two methods for control of Air Pollution.	4
	OR	-4
(b) Enumerate on any two approaches adopted in secondary treatment of waste wa	iter. 4
(c)	Discuss the causes and effects of noise pollution.	3
Q. 4 (a)	Discuss the salient features of Forest (Conservation Act), 1980. OR	4
(a)	Enumerate the functions of State Pollution Control Board.	4
(b)	What are the agents responsible for Ozone layer depletion? What are the effect Ozone depletion.	s of 4
(c)	What are the major impacts of acid rain and how can we control it?	3
Q. 5 (a)	Explain the concept of carbon credit.	4
(b)	What is EIA? Explain the procedure for obtaining Environmental Clearance.	4

TURN OVER

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2

(c)	, and the state of	3
	sustainability?	
	OR	
(c)	Explain the conservation of water using the 3R approach.	
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Q. 6 (a)	Explain the concept of 'Green Buildings'.	4
(b)	Explain the role of technology in Environment and health.	4
	OR	
(b)	Explain disaster management with respect to Tsunami.	4
(c)	Give an account of indoor air pollution.	3
	그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그 그	

SCHOOL OF PAHARMACY SEMESTER I EXAM 2016

First Year B.Pharm Total Marks: 80 Date: 05/05/2017

Sub: Communication Skills & Ethics Duration: 3 Hours

Q.1] Do as Directed:		[403.6
1. They are Knocking down the ald - 1 1 (g)		[10M]
 They are Knocking down the old school.(Change the voice) Harry planted a tree.(Identify the tense) 		
of the flock of birds (was were) flying south contact	lising corre	ot voul
 4. The party start before the show over.(correct the sentence) 5. i live In john street in Norwich England.(punctuate) 6. It has become the rellection 	using corre	ct verb)
of It has occome the fall ving point for	(4.11	
7. In 1956, the hamals went on their first ever strike demanding decer (Change into Infinitive)	(Add questi it wages	on tag)
8. The horse is excellent in swimming		
(Frame a Wh-type question to get the underly)	
9. Give Synonym: Obedient 10. Give Antonym: Knowledge)	
-		
2.2] Fill in the blanks:		
		[5M]
1. Almost all politicians were involved the scandal.		•
 I am looking forward having a meeting with you next week. It doesn't matter whether they get the job not. 	=	_
4. We played very well we still lost the		,
5. The book was the table.	_	
		-
3] Give the Phonetic Transcription:		
ranscription:		[10M]
1. Dancing 6. Theory		
2. Lound		
7. Pet 8. Form		
5 Voc. 9. Art		
10. Home		

Q.4] Define the following: { Any Five }

[10M]

- 1. Communication
- 2. Gesture
- 3. Technical Communication
- 4. Resume
- 5. Leadership
- 6. Stress Management

Q.5] Write Short Notes on: { Any Three }

[15M]

- 1. Barriers of Communication
- 2. Five C's of Technical Communication
- 3. Types of Letters
- 4. Report Writing
- 5. Flow of Communication in Organization
- 6. Impact of Media & Technology
- 7. Family Responsibilities

Q.6] Write an ESSAY on any one of the following topics:

[10M]

- 1. "Pharmacy Day", celebration in the campus
- 2. "Field Project" that you did for Communication Skills Subject
- 3. "Health is Wealth" Discuss
- 4. Discuss how you will Communicate with Cancer Patient & Mentally Challenge
- Q.7] Drift a Report for a General Meeting of all the co-ordinators for keeping the events For the upcoming fest in the college. [10M]

Q.8] Attempt any One:

[10M]

(a) You want to go abroad for your Higher studies in Pharmacy. Write an Inquiry Letter to Citibank inquiring about an educational loan.

OR

(b) You are dissatisfied with the garments consignment you have received from a supplier. Write down a Complaint Letter for the same and mention the details regarding the problem.

*****BEST OF LUCK*****