Q. P. Code: 35625

(3 Hours)

Total Marks: 70

N.B.: 1. All questions are compulsory

2. Figures to right indicate full marks

Q1) A] Answer the following questions

- a) Explain terms: Dihedral angle, Conformation, Ring Flipping, 1,3-diaxial interaction (4M)
- b) Draw all possible resonating structures of Anthracene and Phenenanthrene
- (c) Give identification test for

(2M) (3 M)

- i) An alcohol ii) Carboxylic acid iii) Aromatic primary amine
- C] Give the products for the following reactions (Any six)

(6 M)

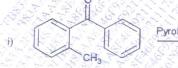
- Strong NaOH i) p-Nitrobenzaldehyde
- ii) Anthracene
- iii) 2 Moles of Benzaldehyde Aq. alc. KCN
- iv) Ethyl benzoate
- v) CH₃CH₂COCI+ t-BuOH -
- vi) C6H5NH2+C6H5COCI-

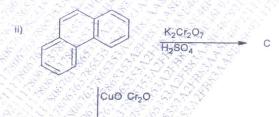
- vii) CH₃COOH i) HN₃ ii) H₂O
- Q2) A] Give the mechanism of any two rearrangements of the following
- (4 M)

- i) Steven alkylation
- ii) Favorski rearrangement
- iii) Hoffman rearrangement
 - B] Complete the following reactions

- (3 M)
- ii) $C_6H_5COCH_3$ $\frac{CF_3CO_3H}{CHCI_8}$ Raney Ni/H₂ i) C₆H₅CH=CHCH₂OH -
- iii) C₆H₅COCI NaBH₄
- C] Identify A, B, C and D

(4 M)





Q3)A] Draw conformers of n-butane and arrange them in the order of relative stability (2 M)

Page 1 of 2

B] Draw neatly the cis and trans conformers of cyclohexane-1,2-diol and briefly dis	cuss stability
for them	(3 M)
C] Attempt the following conversions (Any three)	(6 M)
i) Benzoic acid to phenylacetic acid ii) Salicylic acid to catechol	
iii) o—Methylbenzophenone to anthracene iv) Phenol to ethylphenyl ether	
m) 6—Methyloenzophenone to anumacene 11) I henor to cary party.	
Q4) A] Discuss any two methods of preparation of carboxylic acid	(4 M)
B] Write structure of products formed	
i) When naphthalene is reacted with	(3 M)
	loride in cold
ii) Which product gets formed when beta-napthol reacts with benzene diazonium ch	(1 M)
	(3 M)
C] Give detailed mechanism for alkaline hydrolysis of ester	(3 IVI)
Q5) A] Explain electrophilic substitution on phenol with respect to activation	of ring and
orientation. Cite example of nitration and bromination	(4 M)
B] Suggest products for following	(4 M)
CuBi B	(4111)
NaNO, HCI	
C ₆ H ₅ NH ₂ O-5 Deg.C A	
C] A molecule C ₄ H ₈ O ₂ (A) on treatment with thionyl chloride gives C ₄ H ₇ ClO	(B). This on
treatment with ethyl alcohol in presence of H+ gives C5H812O2 (C). C on hydrolysis	reverts back
to product A. Write structures for A, B and C	(3 M)
Q6)A] Give mechanisms for the following (Any 2)	(4 M)
i) Reformatsky reaction ii) Claisen condensation iii) Beckman alky	lation
B] Give any three methods for the preparation of alcohols	(3 M)
C] A hasty chemist forgot to label the containers and now wants to use basic	
chemistry to solve the problem. He has four containers namely A,B, C and D a	
identify which of them contains propionic acid, benzamide, o-toludine and aceto	
843 BM (ACT ACT ACT SEA SIME ALTO SEA A PROPERTY ACT ALL SEA ACT ACT ACT ACT ACT ACT ACT ACT ACT AC	phenone. He
carried out following four reactions to arrive at conclusion:	

Container A: Compound+ NaOH, boil and smell of ammonia

Container B: Added NaHCO₃ to compound and observed a brisk effervescence

Container C: Added 2,4-DNP and observed thick orange precipitate

Container D: Compound in Conc.HCl+ NaNO2 in HCl at 0-5 deg.C, mix and add beta-nappthol in NaOH gave orange dyestuff. Identify Contents of container A,B, C and D (4 M)

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Q. P. Code: 40385

[Time: Three Hours]

i] permangnometry

[Marks: 70]

3M

- 1. NB: Please check whether you have got the right question paper.
- 2. All questions are compulsory
- 3. Figures to right indicate full marks
- 4. Draw neat labelled diagram, write chemical reaction and give example wherever necessary
- 5. Attempt answer of each main question on new page
- Q.1 [A] Explain the following terms 5M i) complexing Agents ii) common ion effect iii) solubility product iv) masking Agent v) Partition coefficient [B] Answer the following Questions 10M i) State faraday's first & second law ii) Balance the following reaction $Cr_2O_7^{-2} + Fe^{+2} + H^+ \rightarrow Cr^{+3} + Fe^{+3}$ iii) What is half wave potential? iv) explain IIKovic equation v) Partition coefficient of solute between water and diethyl ether is 5. If 25 ml of an aqueous solution of a compound is extracted with 15 ml of organic solvent, what percentage of original solute will be found in organic layer after equilibrium? Q.2 (a) Explain in brief different methods of argentometric titration 4M (b) Give descriptive account of Coulometric techniques 4M (c) Explain levelling and differentiating effect of solvent in non aqueous titration 3M Q.3 (a) Enlist the components of Pharmacopoeial monograph for API as per IP and give the principle involved in the assay of Aspirin API 4M(b) what is the difference between iodometry and iodimetry titration 4M (c) write a short note on pulse polarography and give application of polarography 3MQ. 4 (a) Explain Ostwalds theory of neutralization indicator 4M (b) Enlist unit operations in gravimetric analysis and discuss Precipitation in detail 4M
- Q. 5 (a) A sample of drug A was analysed and percentage concentration obtained after analysis was as follows 8.50, 8.75, 8.14,8.20, 8.58. Calculate mean, median, variance and RSD for the given data

iii] iodimetry

(c) Name the analytes assayed by each of the following type of redox reaction

ii] cerrimetry

(b) write short note on following (i)pM Indicators (ii) demasking agents 4M
(c) classify solvent extraction methods and explain any one in detail 3M

Page 1 of 2

Q. P. Code: 40385

Q.6 (a) Discuss construct	ion and working of Oxygen flask combustion method	4M
(b) 50ml of 0.5 M HCl is	titrated with 0.5 M NaOH. Calculate the pH values at the start	of
titration & after addition	of 5,15,25 ml of titrant.	4M
(c) Calculate gravimetric	factor for	3M
Substance sought	Substance weighted	
P	Ag ₃ PO ₄	
Ba	BaSO ₄	
At wt. S=32.06, O	=15.99, P=38.97, Ba=137.33, Ag=107.87, C=12, H=1, O=16, F	e=55.84
	그리고	

Date : 15/5/18

Q. P. Code: 36190

	[Time: 3 hours] [Marks 86	0]
NB:	 All questions are compulsory Figures to the right indicate full marks 	
b.	Classify dispersed systems. Comment on role of surface free energy in suspension stability. Comment on physiological factors affecting skin penetration. Enlist desirable features of suppositories. Give disadvantages of suppositories as dosage forms. Discuss problems/hazards involved in handling of blood products. Enlist the various quality control tests on ligatures. Discuss tensile strength determination of catgut.	3 2 3 2 3
f.	Discuss pharmaceutical applications of emulsions.	2
	Enlist various quality control tests for suspensions. Elaborate with suitable diagram large scale manufacturing of suspensions.	4
	OR Enlist various quality control tests for emulsions. Elaborate on any one equipment used in	
b. c.	emulsion production. Enlist properties of an ideal suppository base. Elaborate on any one suppository base. Explain and classify non-absorbable sutures. Elaborate on any one non-absorbable suture of natural origin.	4 3
	Explain the following (i) DLVO Theory (ii) Schulze Hardy Rule Discuss Plasmapheresis OR	4
	Discuss Gamma Globulin preparations. Enlist various Quality control tests for suppositories. Explain any one test.	3
b. c.	Discuss salient features of raw materials used in semisolid preparations. Classify Emulsifying agents. Explain selection of emulsifier by HLB method. What are plasma volume expanders? Discuss the salient features of plasma volume expanders with suitable examples.	4 4 3
	OR Discuss steps involved in production of clinical grade Dextran in brief.	
b	Enlist various methods for preparation of suspensions. Explain any one method in detail. Explain large scale manufacturing of any one semisolid preparation. Explain salient features of Theobroma oil as a suppository base. OR	4 3 4
	Describe any two large scale manufacturing techniques of suppositories.	
b. 1	Discuss manufacturing and processing of catgut. Elaborate on physical stability of emulsions. OR	3 4
c. (Elaborate on preservation of emulsions. Classify Penetration enhancers with examples. Discuss any two methods to evaluate skin penetration.	4

CBSGS, Sem IV

Microbiology, 17/5/18

Q.P. CODE: 36180

[Time: 3 Hours]

[Total Marks: 70]

Instructions:

All Questions are compulsory. Q.1 Answer the following	
 a. Name the diagnostic test for tuberculosis b. Define and give significance of DRT c. Explain bacterial capsule d. Name the causative agent of typhoid e. Name the biological indicator of filtration sterilization f. Define resolution limit g. Define oncogenic viruses h. Write incubation conditions in sterility testing and name of sterility testing media i. Name any two fungal infections with the causative agent j. Name any two chlamydial infections with the causative agent k. How to prevent contamination in an aseptic area 	
Q.2 a) Explain fluorescence microscopy using a neat labelled diagram with its applications	4
b) Discuss moist heat sterilization with respect to method, the mechanism of action and applications	2
c) Write a note on preservation of bacteria	3
Q.3 a) Describe lysogenic cycle for bacteriophage	4
b) Discuss in detail kelseyskeys test of disinfectant evaluation	4
c) Distinguish between bacteria and fungi	3
OR	
Distinguish between prokaryote and Eukaryote	
Q.4 a) Enlist different methods of counting of bacteria and explain any one viable method	4
b) Discuss in detail infections caused by Protozoa	4
c) Explain asexual methods of fungal reproduction	3
Q.5 a) Discuss in detail radiation sterilization or gaseous sterilization	4
b) Write a note on alcohol as a disinfectant	4
c) Explain economical significance of algae	3
0.6 a) write a note on gram positive cell wall using a neat labelled diagram	4
b) Explain phases of bacterial growth cycle	4
c) Explain the principle of Acid fast staining technique	3

5. Y. B Phanm (Sem II) (BSGS - P' cology 1.

Q.P. Code:04478

		[Time: 2½ Hours]	arks:
6		Please check whether you have got the right question paper.	
		N.B: 1. All questions are compulsory.	
		2. Figure to the right indicate full marks.	
1.	(a)	Answer the following	1
		Define bioavailable and bioequivalence	
	11.	Differentiate between potency and efficacy of a drug	
	111.	What are the clinical uses of skeletal muscle relaxants?	
	IV.	Classify muscarinic receptors along with their location	
		Carbonic anhydrase inhibitors are used as mild diuretics. Justify	
	VI.	Give the mechanism of action of quinidine.	
1.	(b)		03
		그 그는 그는 그는 그는 그는 그는 그는 그 사람들이 얼마나 하는 것이 되었다면 하는 것이 없었다.	
		What do you mean by pharmacological antagonism? Give examples.	
	11.	Differentiate between sympathetic and parasympathetic nervous system in terms of present neurotransmitters.	
	111.	Give the pathogenesis of cardiac arrhythmia.	
2.		Answer any two of the following	08
		Classify parasympathomimetics and add note on their therapeutic uses.	
		Give the pharmacological action of Acetylcholine.	
	111.	Write note on depolarizing and non-depolarizing ganglionic blockers.	
	(b)	Answer any one of the following	03
	1.	Explain the various phases of drug metabolism with example.	
	II.	Define renal clearance, Volume of distribution and half life of drug	
3.	(2)	Answer any two of the following	08
٥.		Classify antiarrythmic agents and explain the role of calcium channel blockers.	
		Explain the role of diuretics in managements of hypertension	
		Explain various toxic effects of drug on different organs and systems.	
	(b)	Answer any one of the following	03
	1.	Give the pharmacotherapy of Angina pectoris	
		How pharmacokinetic factors affects action of drug?	
4.	(a)	Answer any two of the following	08
	1.	Discuss atropine like substitutes in-detail.	
	ıll.	Enlist the therapeutic uses of Noradrenaline and explain its mechanism of action.	
	III.	What treatment will you prescribe to the patient suffering from cardiac arrhythmia if he/she is having asthma?	
	10 T 128		

Q.P. Code :04478

	(b)	Answer any one of the following	03
	1.	Classify nicotinic receptor antagonist	
	· II.	Explain the role of sympathomimetics in asthma	
5.	(a)		08
	1.	Discuss various types of receptors and explain their functioning with examples.	
	11.	Explain physiological and chemical antagonism with suitable examples.	
	III.	Differentiate between competitive and non-competitive inhibition.	
	(b)	Answer any one of the following	03
	1 2 2	Classify routes of drug administration.	
	L	Drug administered via different routes will have same half life. Justify	
	H.	Drug administered via different routes will have same from the sound	
6.	(a)	Answer any two of the following	08
	1.	Write note on antihyperlipidemic drug	
	11.	Give the classification of sodium channel blocker along with their uses.	
	111.	Give the therapeutic uses of following drugs (any 4)	
		a) Succinly Co A	
		b) Physostigime	
		c) Quinidine	
		d) Spirinolactone	_
		e) Diltiazem	
	(b)	Answer any one of the following	03
	1.	Discuss the effect of protein binding capacity and related drug action.	
	- 11	Discuss nations related factors affecting drug action	