LJ-10200

(2 Hours)

[Total Marks: 75

- **N.B.**:(1) Question No. 1 is compulsory.
 - (2) Solve any four questions from remaining six questions.
 - (3) All questions carry equal marks.
 - (4) Atomic wts: H = 1, C = 12, N = 14, O = 16, Na = 23, Mg = 24, S = 32, C1 = 35.5, Ca = 40.
 - 1. Solve any **five** from the following:—

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- (a) What is galvanic corrosion?
- (b) Write a note on mining of crude petroleum.
- (c) What are the applications of Composite materials.
- (d) List any six principles of green chemistry.
- (e) Explain the carbon membranes.
- (f) Give the composition, properties and uses of Duralumin.
- (g) By Kjeldahl's method, 1·5 g of coal sample was analysed. The ammonia evolved was absorbed in 50 ml of 0·1 N H₂SO₄. After absorption, the excess H₂SO₄ required 35 ml of 0·1 N NaOH for neutralisation. Calculate the percentage of nitrogen.
- 2. (a) 2.5 g of air dried coal sample was taken in a silica crucible. After heating it in an electric oven at 105-110°C for 1 hr, the residue weighed 2.410 g. The residue was heated in a silica crucible covered with a vented lid at a temperature of 950 ± 20° for exactly 7 min. After cooling the weight of residue was found to be 1.78 g. The residue was then ignited at 700-750°C to a constant weight of 0.2469 g. Calculate the percentage of fixed carbon in the sample.
 - (b) Define corrosion. Explain the effect of the following factors on the rate of corrosion.
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- (i) Position of metal in Galvanic series.
- (ii) Relative area of the anodic and cathodic part.
- (c) What is compacting? Explain the powder injection moulding method.
- 3. (a) Write notes on (any two):—

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- (i) Indergranular corrosion
- (ii) Stress corrosion
- (iii) Metal cladding.
- (b) What is bio-diesel? Explain the 'trans-esterification' method used for its production. 5 Give its advantages.
- (c) What is powder metallurgy? How are metal powders prepared? (any 3 methods). 4

- 4. (a) Calculate the weight and volume of air required for complete combustion of 5kg coal with the following compositions: C = 85%, H = 10%, O = 5%.
 - (b) Define composite materials. Write a note on fibre reinforced composites. 5
 - (c) Calculate the percentage economy for the following reaction. Synthesis of Maleic Anhydride by oxidation of benzene:—

- 5. (a) Define Catalyst. What are the types of Catalyst? Discuss the characteristics of an ideal catalyst (any three).
 - (b) Define paints. Disucss the functions and examples of the constituents of paints 5 (any two).
 - (c) Write a note on non-oxide powder, Silicon Carbide.
- 6. (a) Give the conventional? Green chemistry route for production of Adipic acid. 6 Highlight the green chemistry principle addressed in this case.
 - (b) What is Catalysis? Give the various types with appropriate examples. 5

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- (c) What is cathodic Production? Discuss the sacrificial anodic protection method.
- 7. (a) Explain the adsorption theory of heterogenous catalysis. 6
 - (b) Define cracking. What are the types of cracking methods used? Differentiate 5 between the tuib.
 - (c) Write a note on structural composites.