

Applied Chemistry FE-Sem II

May 2013 22/05/13

88: 1ST HALF-13 (r)-JP

Con. 6921-13.

(REVISED COURSE)

A.C.II

GS-5481

(2 Hours)

[ Total Marks: 60

N.B. (1) Question No. 1 is compulsory.

- (2) Attempt any **three** from remaining **five** questions.
- (3) All questions carries equal marks.
- (4) Atomic weight :—

$$H = 1,$$

C1 = 35.5

$$C = 12$$

Ba = 
$$137.3$$
,

$$N = 14,$$

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  $Mg = 24,$   $O = 16,$   $Na = 23,$ 

$$S = 32$$
,

$$Ca = 40$$

1. Answer any **five** from the following:—

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- (a) Why silver, gold and platinum do not undergo oxidation corrosion?
- (b) Define Octane number and Cetane number. Give their significance.
- (c) Give the composition, properties and uses of German silver.
- (d) Give classification of composite material.
- (e) What is Green chemistry? List the 12 principles of Green chemistry.
- (f) State the characteristics of a good paints.
- (g) A coal sample was subjected to ultimate analysis, 0.6 gm of coal on combustion in a Bomb colorimeter, produces 0.05 gm BaSO<sub>4</sub>. Calculate the percentage of 'S' in coal sample.
- 2. (a) What are metallic coatings? Distinguish between Galvanizing and Tinning.
- 6

(b) Explain refining of petroleum with suitable diagram.

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(c) Calcualte % atom economy for following reactions:—

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- 3. (a) A coal sample has the following composition by weights : C = 82%, H = 3%, O = 8%, S = 2%, N = 2% and Ash = 3%. Calculate the minimum amount of air required both by weight and volume for complete combustion of 2 kg of coal. (mol-wt. of air = 28.949 gm). 5
  - (b) Explain traditional and greener route of production of Indigo dye. By this reactions which principle of green chemistry is shown?
  - (c) How is the rate of corrosion influenced by:

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- (i) pH of medium
- (ii) Relative area of cathode and anode parts?

(iii) Cobalt

(iv) Molybdenum(v) Tungsten.

## Con. 6921-GS-5481-13. 2 6 4. (a) Write a note on Compacting and Sintering. 5 (b) Explain wet corrosion in acidic medium with schematic diagram and mechanism. (c) Explain Laminar composite with suitable diagram. Example. 4 6 5. (a) What is bio-diesel? Explain the method to obtain bio-diesel from vegetable oil. Give advantages of bio-diesel as a fuel. 5 (b) Distinguish between Brass and Bronze. (c) State the chemical factors influencing adhesive action. 4 6. (a) What is cathodic protection? Describe impressed current method of corrosion control. 5 (b) A gaseous fuel has the following composition by volume: $H_2 = 10\%$ , $CH_4 = 30\%$ , $C_3H_8 = 20\%$ , CO = 20%, $CO_2 = 15\%$ , $N_2 = 5\%$ . Calculate the volume of air required for complete combustion of 1m<sup>3</sup> of this gas. 5 (c) Explain the effect of following elements on alloying:— (i) Nickel (ii) Chromium