

- 4(a) Explain hydrodynamic and Thermal Boundary Layer 5  
(b) With the help of dimensional analysis method, prove that for convection:-  
$$Nu = \text{constant} \times (Re)^m \times (Pr)^n$$
 8  
(c) Explain shape factor and its properties. Find the shape factor of a cylindrical cavity of diameter  $D$  and depth  $H$  w.r.t. itself. 7
- 5(a) A hot cylinder ingot of 60 mm diameter and 200 mm long is taken out from the furnace at  $900^\circ\text{C}$  and dipped in water till its temperature falls to  $500^\circ\text{C}$ . Then it is directly exposed to air till its temp falls to  $100^\circ\text{C}$ . Find the total time required for the ingot to reach the temperature from  $900^\circ\text{C}$  to  $100^\circ\text{C}$ . Take the following properties  $K=50 \text{ W/m}^\circ\text{C}$ ,  $C$  (specific heat of ingot)  $=200 \text{ J/m}^\circ\text{C}$ ,  $\rho$  (density of ingot)  $=800 \text{ kg/m}^3$ ,  $h_w$  (heat transfer coefficient in water  $=200 \text{ W/m}^2 \text{ }^\circ\text{C}$ ,  $h_a$  (heat transfer coefficient in air)  $=20 \text{ W/m}^2 \text{ }^\circ\text{C}$ , temperature of air or water as  $30^\circ\text{C}$ . 8
- (b) Write short notes on: 12  
(1) Boiling Curve  
(2) Shape Factor Algebra  
(3) Critical Radius and Thickness of insulation.
- 6(a) Explain the terms: Effectiveness, NTU, and LMTD 6  
(b) What are Fourier and Biot numbers? What is the physical significance of these numbers? 8  
(c) Distinguish between Specular and diffuse radiation 6
- 7(a) A distillation column containing a mixture of benzene and toluene is at pressure of 1 bar and temperature of  $105^\circ\text{C}$ . The liquid vapor phase contain 20 mol% of benzene and 55 mol% of toluene respectively. At  $105^\circ\text{C}$  vapor pressure of the toluene is 0.72 bar and its diffusivity is  $5.2 \times 10^{-6} \text{ m}^2/\text{s}$ . Assuming the equimolar diffusion, calculate the molar diffusion flux of toluene if the diffusion zone is 0.35 m thick. Take universal gas constant as  $8.314 \text{ kJ/kg mol K}$ . 12  
(b) State and explain Fick's law of diffusion and derive the condition for equimolar diffusion. 08