

QP Code :11836

(OLD COURSE)
(2 Hours)

[Total Marks: 75]

- N.B. :**(1) Question no. 1 is compulsory.
 (2) Attempt any four questions from the remaining six.
 (3) Figures to the right indicate full marks.
 (4) Atomic weights H=1, C=12, N=14, O=16, Na=23, Mg=24, S=32, Cl=35.5, Ca=40

1. Answer any five questions from the following:— 15
- (a) 1 gm of coal sample was used for determination of nitrogen by Kjeldahl's method. The ammonia evolved was passed into 50 ml of 0.1N H₂SO₄. The excess acid required 40ml of 0.1N NaOH for neutralization, Calculate the percentage of nitrogen.
- (b) Give the composition, properties and uses of tinman's solder.
- (c) Write are the characteristics properties of a good paint.
- (d) Write the classification of composite materials.
- (e) Write a short note on green fuel.
- (f) What is catalysis? Explain different types of catalysis with one example each.
2. 6
- (a) Explain refining of petroleum. 5
- (b) Define corrosion. Explain interanular corrosion with appropriate diagram and examples. 4
- (c) Explain the effect of the following alloying elements on steel 4
- (i) Cr (ii) Ni (iii) Mo (iv) W
3. 6
- (a) Calculate the weight and volume required for complete combustion of 1 kg of coal containing C= 60%, H= 5%, O=7%, N= 3% and remaining being ash.(M.W of air =28.949) 5
- (b) Define cracking. Discuss fluid bed catalytic cracking. 4
- (c) Explain the following factors affecting the rate of corrosion 4
- (i) pH of medium
- (ii) position of metal in galvanic series.
4. 6
- (a) Explain the production of alcohol from molasses. 5
- (b) Write a short note on fibre reinforced composites. 4
- (c) Calculate the percentage atom economy of the following reaction 4
- $$\text{C}_6\text{H}_6 + 4.5\text{O}_2 \xrightarrow{\text{V}_2\text{O}_5} \text{C}_4\text{H}_2\text{O}_3 + 2\text{CO}_2 + 2\text{H}_2\text{O}$$