FESANI COBGS) AC-II

25/5/15

(REVISED COURSE) Q.P. Code: 1049

(2 Hours) [Total Marks: 60

TURN OVER

1	.D	(1) Question 140. I is Compulsory.	No.
		(2) Attempt any three from remaining fix questions	Surrey .
		(3) All questions carry equal marks.	
		(4) Figures to the right indicate full marks.	
		(5) Atomic weights: H=1, C=12, S=32, N=14, O=16, Cl=35.5, Ba=137.3,	
		Na=23, Mg=24.	
		and the second s	
1.	Answ	ver any five from the following:-	15
	(a)	What are propellants? State importants characteristics of good propellant.	10
	(b)	Compare Galvanizing and Tinning.	
	(c)	Give composition, properties and uses of Wood's Metal.	
	(d)	Write a note on 'Green Reagent'.	
	(e)	Define terms :-	
		(i) Composite material (ii) Matrix phase (iii) Dispersed phase.	
	(f)	List three main constituents of Varnish & give functions of each.	
	(g)	A coal sample was subjected to ultimate analysis:	
		1.6 gm of coal on combustin in a Bomb calorimeter gave 0.47 gm of BaSO4	
		Calculate % of sulphur in the coal sample.	
2.	(a)	What is dry corrosion? Explain with example how nature of oxidised product	6
		affect the rate of corrosion.	
	(b)	What is cracking? Explain fixed bed catalytic cracking with diagram.	5
	(c)	Calculate percentage atom economy for the following reaction w.r. to methy	4
		iso-cyanate	
		$CH_3NH_2 + COCl_2 \rightarrow CH_3N + COCl_2 \rightarrow C$	40
		methyl iso cyanate.	
2	(0)	A gagaging final handle following commonition by well-	
3.	(a)	A gaseaus fuel has the following composition by volume. CH = 35% CH = 5% CO = 15% H = 40% N = 1 years year aver = 40%	6
		$CH_4 = 35\%$, $C_2H_4 = 5\%$, $CO = 15\%$, $H_2 = 40\%$ $N_2 = 1$ water vapour = 4% Calculate volume & weight of air required for complete combustin of $1m^3$ of fuel	
		[mol.wt of air = 28.94]	
	(b)	Explain conventional & green synthesis of adipic acid. Mention the green	_
	(0)	chemistry principle involved.	5
	(c)	How the rate of corrosion influenced by following factors.	4
	()	(i) PH of medium (ii) Over voltage.	4
		(ii) overvoitage.	
4.	(a)	What is powder Metallurgy? How are metal powders prepared using.	6
	harry)	(i) Atomization (ii) Chemical reduction	U
4	(b)	What is cathodic protection? Explain Impressed current method of corrosion	5

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Write a note on 'Sandwitch panel' type layered composites. 5. (a) What is Bio-diesel? Explain the trans esterification method for its synthesis. 6 Mention advantages of biodiesel as fuel. (b) What are alloys? Explain any four purposes of making alloys with suitable example 5 Discuss the physical factors influencing adhesive action. (c) 6. (a) Write a note on differential aeration corrosion. 2.5 gm of air dried coal sample was taken in a silica crucible, after heating it in an 5 (b) electric oven at 110°C for 1hr the residue was weighed 2.41 gm. The residue was heated in Silica crucible covered with vented lid at a temperature 925 ± 25°C for exactly 7 minutes. After cooling the weight of residue was found to contain 1.98 gm. The residue was then ignited to a constant weight of 0.246 gm. Report the results of above analysis. (c) Explain the effects of following elements on alloying: 5 (i) Nickel (ii) Chromium Cobalt (iii) Molybdenum (iv) Carbon. (v)