Q.P. Code: 21729

TURN OVER

[Total Marks: 70 (3 Hours) N.B.: (1) All questions are compulsory. 1. (a) Answer the following. (i) State the type of strains present in cyclopropane. (ii) Justify "Anthracene is more stable than napthalene". (iii) m-Nitrophenol is less acidic than p-nitrophenol. 2 (b) Give distinguishing test for the following: (i) Aniline and dimethylamine. (ii) Acetophenone and benzophenone. (c) Complete the following reaciotns (Any 10). 10 (i) Naphthalene  $\frac{O_2, V_2 O_5}{450 - 500^{\circ}C}$ (ii) Phenol +  $CHCl_3 \xrightarrow{OH^{\theta}}$ (iii) Cyclopentanone  $\frac{NH_2 - NH_2}{base}$ (iv)  $H_5 C_2 COCl + t - BuOH$ (v) Acetone  $\frac{Ph_3P = CH(CH_3)}{}$ (vi)  $C_6 H_5 CH_2 Br + C_2 H_5 OH \xrightarrow{Na} \rightarrow$ Acetophenone  $\xrightarrow{\text{i)} \text{NH}_2\text{OH}}$ Phthalic acid 200°C (viii) Ethyl benzoate aq. NaOH Acetaldehyde + CH<sub>3</sub>NO<sub>2</sub> OH<sup>6</sup> xi) Ethyl adipate NaOCH<sub>3</sub> > 2. (a) Attempt the following coversions (any two) (i) 2 - Cholrocyclohexanone to ethyl cyclopentane carboxylate. (ii) 4 - Chlorobenzaldehyde to 4 - Chlorocinnamic acid.

LO-Con. 416-15.

(iii) 3 - Methylaniline to 3 - Methylbenzoic acid.

## Q.P. Code 21729

2

- (b) Write products formed when
  - (i) Ethylene oxide reacts with phenylmagnesium bromide.
  - (ii) Anisole heated with 57% HI
  - (ii) Propanone treated with Zn (Hg)/HCl
- (c) A molecule C<sub>4</sub> H<sub>8</sub> O<sub>2</sub> (A) on treatment with thionyl chloride gives C<sub>4</sub>H<sub>7</sub>ClO (B) This on treatment with CH<sub>2</sub> N<sub>2</sub> and solid Ag<sub>2</sub>O gives an intermediate ketene (C) Which on hydrolysis gives C<sub>5</sub> H<sub>10</sub> O<sub>2</sub> (D) Deduce A, B, C, & D. (Note: Compound A gives strong efferevesence with sodium bicarbonate solution)
- 3. (a) Complete the reations with relevant mechanism (any two)

(i) 3 - Cyclopropyl - 2 - phenyl - butane - 2, 3, - diol 
$$\xrightarrow{H^{\oplus}}$$

- (ii) Acetone + diethyl succinate  $\xrightarrow{\theta}$  Ot -Bu  $\Rightarrow$
- (iii)  $H_5 C_2 O CO O C H_5 + C_6 H_5 C H_2 C O C_2 H_5 \xrightarrow{O C_2 H_5}$
- (b) Answer the following
  - (i) Discuss the optical activity for cis and trans 1,3-dimethylcyclohexane.
  - (ii) Cis-cyclohexane- 1, 4-diol predominantly exist in the boat or twist boat form Justify.
- (c) Complete the following.

- (ii) Ethylbenzene Na/Liq NH<sub>3</sub>
- (iii) Ethyl benzeate NaBH<sub>4</sub>
- 4. (a) Write the products formed.

(i) Authracene 
$$\frac{\text{HNO}_3}{\text{Ac}_2\text{O}} > \text{A} + \text{B}$$

- (ii) Phenanthrene +  $H_3$  C COCl  $\frac{AlCl_3}{0^0C}$   $\Rightarrow$
- (iii) Naphthalene +  $H_2 SO_4 \xrightarrow{160^{\circ}C}$

TURN OVER

3

(b) Give mechanisms for any two of the following with suitable examples.

- (i) Knovanagel condensation
- (ii) Benzil-benzilic acid rearrangement.
- (iii) Acid Hydrolysis of esters.
- (iv) Crossed aldol condensation.
- (c) From the following scheme, identify A, B, C & D.

1 - NaNoza Hel

3. (a) Attempt any two of the following coversions.

- (i) Malonic acid to phenylacetic acid
- (ii) Acetophenone to 2-hydroxy-2-phenylpropanoic acid.
- (iii) t Butyl alcohol to trimethylacetic acid
- (b) Account for the reaction mechanism and stereochemistry for the following reations.
  - (i) (S) 2 Phenylpropionamide on action with OBr gives
    - (S) 1- Phenylethylamine.
  - (ii) 2 Pentyltrimethylammonium hydroxide on action of sodium ethoxide gives 1-pentene (96%) and 2 - pentene (4%) as products.
- (c) Give the products of following reations.

3

- (i)  $C_6 H_5 CHO + NH_3 \xrightarrow{H_2/Ni}$
- (ii) (CH, COC), Ca heat
- (iii)  $C_6 H_5 CH_2 COO CH_2 C_6 H_5 + excess C_2 H_5 OH H^{\oplus}$
- 6. (a) (i) Explain Gomberg reation by giving its mechanism and state any one application for the same.
  - (ii) Suggest suitable reagent for conversion of H, CCH = CH - CHO to CH, CH = CH - CH, OH

1

(b) (i) Discuss any two pathways to synthesize 4 - Methoxy aniline from 4 - Methoxy benzoic acid.

TURN OVER

LO-Con. 416-15.

## Q.P. Code: 21729

4

(ii) Write the products:

Phenyl propanoate 

AlCl<sub>3</sub>

CS,

(b) (i) Give the products and mechanism for reation of 4 - nitrobenzaldehyde with 50% NaOH.

(ii) Complete the following scheme

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Pthalie anhydnide

KOH, 275BV

A NHZ-MZ

Btc