	ii\@A	11 × 1 ×	O-KIKIO	XVIT-INCC	07-12
	B.F	pho	m / som -A	# 1 CBCS -	
6			Sub 1- Gene	I-CBCS- ral Chemistry	Regula.
0					Q. P. Code: 22580
				(3 Hours)	Total Marks: 80
	N.B.:	1) A	All questions are compulsor	у	
		2) A	Answer all sub questions to	gether	
		3) F	figures to right indicate full	marks	
	Q.1	(2)	Explain the terms (Any 5)		5
	Q.1	i) N			5
		,	Absorption dose	A	
			Systemic antacid		
			Astringents		
			Complexing agents	AR PARTY TEC.	
			Cathartics	* * * * * * * * * * * * * * * * * * *	
			KALIOG	ARCHI S	7,
		b) A	answer the following (Any	v 5)	10
				ive effect? In which of the f	ollowing examples inductive
		effe	ct observed: Acetone, Hydr	ogen, Ethane, Flurobutane	**
		ii) [Draw an energy profile diag	gram showing Activated comp	plex, intermediate, activation
		ener	gy and reaction endotherm		RATIO
		iii) I	Explain the term catalysts a	nd classify them	A
			What is mEqv scale? What		70
		v) S	tate and explain significanc	e of emetics as an inorganic p	harmaceutical agent
		vi) I	Discuss the role of Iron as ea	ssential and trace elements.	N. S.
		c) M	latch the following		5
		-)		Column B	
		i)	Column A Selenium	MIMP Extracellular ion	
		ii)	Hypophosphorous acid	Antioxidant	
		iii)	Chloride	Essential & trace	e element
		iv)	Potassium	Acidifying agen	ts
		v)	Dil. Hydrochloric acid	Intracellular ion	
	Q.2	a) W	rite a detail note on Kinetic	Elsotope Effect	4 =
		b) A	nswer the following (Any	2)	4
				of action for Hydrogen Perox	vide and
			pitated sulfur	42 motion for frydrogon f croz	nav and
		_		te any two examples for the s	ame
			*	, , , , , , , , , , , , , , , , , , , ,	

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Q. P. Code: 22580

	iii) Give the role of iod	ine as essential ion and enlist products for the same.	
	c) Enlist various units of d) Write down role and calcium as major physical	of radioactivity. Give the properties of gamma particles. 2 I conditions related to changes in the concentration of sodium ological ions	
2.3	b) What are the ideal prantacids.c) Add a note on Quant	charge on underlined atom (Any 2)	4 ts of 4 2 2
Q.4	a) Fill in the blanks wai) In BCl ₃ the central at	om is and shows type of hybridization. cronic configuration is and excited state with bond angle of	4
	c) i) Explain reactivity ii) '1,3-Butadiene on ad	topical agents and write details of any two agents vs selectivity principle Idition of HBr at lower temperature gives 1,2 addition emperature 1,4 addition product'. Explain the underlying	4 2
		names of two categories of products formed in it	2
2.5	a) State true or false		4
	· · · · · · · · · · · · · · · · · · ·	gativity of CF3 is higher than CH3 ne atom attached to methane to CCl4 lower is the	
	iii) Bond lengths for hybridized bond	r sp3 hybridized bond is lower than that of sp	
		wn in the column polarizability increases	
	with acidity functions the c) Write a note on cyanic	asted Acid Base catalysis. Explain correlation of reaction rates arough relevant examples. ide poisoning and its treatment tance of calcium and phosphate ions	4 2 2

Q. P. Code: 22580

Q.6 Answer the following (Any 6)

12

- i) Complete the following: 20 millicurie = ---- Bq $6.8 \times 10^{-12} \text{ dps} = --- \text{ Curie}$
- ii) Pd^{100} has a half life of 3.6 days. If its initial amount has $6.02x10^{23}$ atoms, how many atoms would be present after 23 days.
- iii) Enlist electrolytes used in replacement of sodium, potassium and calcium and mention their use
- iv) How many mEq of Magnesium ions are present in 1% MgCl₂ solution
- v) Explain the terms reaction order and rate law
- vi) State and explain Hammond's Postulate

vii) Classify the following agents: silver nitrate, magnesium trisilicate, EDTA, sodium hypochlorite



Q.P. Code:05093

[Time: - 3 Hours]

[Marks: 80]

Please check whether you have got the right question paper.

N.B:

- 1. All questions are compulsory.
- 2. Figures to the right indicate full marks.

1.	a)	Classify dosage forms.	4
	b)	Define i) Compounding and ii) Dispensing	2
	c)	Write in brief about dispensing of proprietary medicines.	2
	d)	Calculate the dose for a 8 year old patient if the adult dose of the same drug is 250 mg.	2
	e)	Define emulsion. Classify emulsifying agents.	2
	f)	Enlist the advantages and disadvantages of powders as a dosage form.	2
	g)	Enumerate the advantages and disadvantages of pills.	2
	h)	Give anyone example of chemical incompatibility. How would you overcome the same?	2
	i)	Enlist the principles of pharmaceutical care	2
	j)	Give any 4 examples of OTC products.	2
		MANOGO CHELLEN WAY	
2.	a)	Enlist the various routes of administration. Explain the oral route in detail.	4
	b)	Highlight the container-closure and labelling directions for (any TWO).	4
		i. Emulsion ii. Suppository iii. Lozenge	
	c)	Comment on the following prescription.	4
		Rx E	
		Calciferol solution - 0.15 ml	
		Glycerin - 0.3 ml	
		Water (q.s.) to - 5 m	
		Label: 5 ml to be taken daily	
3.	a)	Define a prescription and explain its various parts.	4
	b)	Classify ointment bases. Describe any ONE base in detail.	4
		OR	
		Describe in detail the disadvantages of cocoa butter as a suppository base.	
	c)	Discuss compounding and dispensing of effervescent granules in detail.	4
		A DILLON	
1.	a)	How would you formulate a suspension containing an indiffusible solid?	4
		OR	
		Classify solutions. Describe in brief any TWO solutions meant for oral use	
	b)	Discuss the compounding and dispensing of cocoa butter suppositories.	4
	c)	Discuss the responsibilities of a community pharmacist.	4
	a)	Define "health" as per WHO. Discuss the areas in which a pharmacist can get involved to	4
		promote health of a community.	

TURN OVER

6.

Q.P. Code:05093

2

b)	What is patient counselling? Give its significance. Describe in brief the steps involved in patient counselling.	4
c)	Paraphrase the code of ethics for a pharmacist in relation to his trade as laid down by the Pharmacy Council of India.	ı
a)	How many ml of 1: 5000 w/v solution of preservative benzalkonium chloride can be made from 125 ml of 0.2% solution.	4
	OR	
	In what proportions should ointments containing 25% w/w sulphur, 15% w/w sulphur and 5% w/w sulphur be mixed to produce an ointment containing 20% w/w sulphur	4
(b)	Comment on the following prescription.	4
	Rx Hyoscine hydrobromide 200 µg Lactose qs Send 4 packets of divided powders. Label: One powder to be taken 30 mins prior to journey	
c)	i. What is a balanced diet? Give its significance	7
<i>C)</i>	Give the causative agents for the following diseases:	2
	i. Diphtheria	
	ii. Typhoid	
	iii. Scabies	
	iv. Tetanus *	

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[Time: Three Hours]

Q.P. Code:07532

[Marks:80]

		Please check whether you have got the right question paper.	
		N.B: 1. All questions are compulsory.	
		Figures to the right indicate full marks.	
		3. Draw neat, labeled diagrams wherever necessary.	
Q.	a) Answe	er the following.	16
	i.	Justify -blood as a connective tissue	
	ii.	Justify – Pseudostratified epithelium as epithelial tissue	
	iii.	Justify – Bloodgroup AB is universal system.	
	iv.	Justify – Skin as a part of immune system.	
	V.	State the difference between primary and secondary lymphatic organs.	
	vi.	Define –negative feedback mechanism.	
	Vii	Enlist the components of Blood plasma.	
	viii	i. Enlist types of muscular tissue	
		*ANN	04
	b) Answe	r the following:	
	i.	A healthy adult has aboutmillion red blood cells per microliter of blood.	
	ii.	Leukemia is represented by abnormal production of	
	iii.	Als a region where communication occurs between two neurons, or between a	
		neuron and a target cell.	
	iv.	Name two types of adaptive immunity.	
Q.2	a) Answer	any two of the following.	80
	i.	Write a note on composition of blood and its functions.	
	ii.	Describe the process blood coagulation.	
	iii.	Discuss erythroblastosis fetails.	
	b) Write a	short note on any one of the following	04
	i.	Leucopenia and Leukocytosis Anemia and its types	
	ii.	Anemia and its types / //////////////////////////////////	
Q.3	al Answer	any two of the following	08
۷.5	i.	Give detailed account of skeletal muscle contraction	00
	ii.	Write a note on energy metabolism in skeletal muscle.	
	iii.	Describe sliding filament mechanism of skeletal muscle contraction	
		source sharing marrierit meditaliisti of skeletal musele contraction	
	b) Answer	any one of the following	04
	i.	Write a note on neuromuscular transmission	
	ii.	Draw a neat, labeled diagram showing organization of skeletal muscle	

Q.P. Code :07532

Q.4	a) Answer	any two of the following.	04
	į.	Draw a neat, labeled diagram of spleen. Discuss functions of spleen.	
	ii.	Discuss anatomy and functions of lymph node.	
	b) Write a	short note on any one of the following	04
	i.	Myasthenia Gravis	
	ii.	AIDS	
	c) Answer	any one of the following	04
	i.	Describe bone as a connective tissue, its classification and function.	
	ii.	Draw a neat labeled diagram of a neuron.	
Q.5	a) Answer a	any one of the following	04
	i.	Discuss fluid mosaic model of plasma membrane	
	ii.	Explain active transport mechanism	-
	b) Answer a	ny one of the following	04
	i.	What are different types of muscle contraction? Explain with the help of examples	
	ii.	Write a note on functions and properties of muscles.	
			04
	c) Answer a	ny one of the following.	
	i.	Describe the mechanisms involved in migration of WBC	
	ii.	Write a note on process of chronic inflammation.	
Q.6	a) Answer ar	ny two of the following	08
	i.	Write a short note on synthesis of hemoglobin	
	ii.	Explain the process erythropoiesis in detail.	
	iii.	Explain different types of hypersensitivity reactions.	
			04
	b) Answer an	y one of the following	
	j.	Describe the process of chemotaxis and its importance.	F
	ii.	Discuss different cells involve in immune response.	



Correction: T9831 / T2511 - Anatomy.Physiology &Pathophysiology -I QP Code: 07532

University of Mumbai <support@muapps.in>



Q.P. Code: 24793

[Time: 3 Hours]

[Marks: 80]

Please check whether you have got the right question paper.

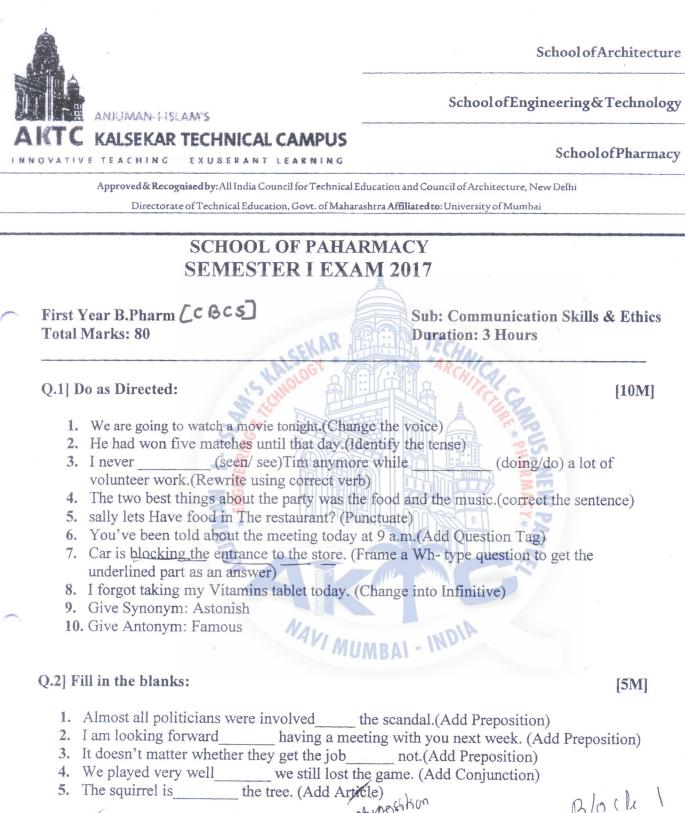
1. All questions are compulsory. N.B:

1.	a)	Draw the structure of α - D mannose by using Haworth projection formula	1
	b)	Draw the structure of D-xylose by using Fischer projection formula	1
	c)	Name proteolytic enzyme required for digesion of protein	1
	d)	Explain metabolism with example	1
	e)	Enlist fat soluble vitamins	1
	f)	Define mutarotation	1
	g)	Give the structure of coenzyme of Vitamin B ₂	1
	h)	Name the pyrimidine nitrogenous bases	1
	i)	Write the structure of C ₄ epimer of glucose	1
	j)	Draw the structure of $18:2(\Delta^{9,12})$	1
	k)	Draw the structure of ATP	1
	1)	Deficiency of Vitamin-A leads to	1
	m)	Give the name and draw the structure of basic amino acids	2
	n)	Differentiate sucrose and maltose	2
	0)	Explain the a nein strategic of proteins	2
	p)	Enlist essential amino acids//4// MUMRAI - INDIA	2
		MUMBAI - IND	
2.	a)	Explain the primary and tertiary structure of proteins	3
	b)	Explain ATP as energy carrier	3
	c)	Write a note on Vitamin-B ₆ or Vitamin-B ₇	3
	d)	Explain nucleotides	2
	e)	Enumerate silent features of digestion of proteins	1

Q.P. Code: 24793

2

3.	a)	Write a note on starch	3
	b)	Differentiate DNA and RNA	3
	c)	Write a note on Vitamin-C	3
	d)	Explain the relationship between standard free energy change and equilibrium constant	2
	e)	Comment on conversion of glucose to energy in RBCs	1
4.	a)	Classify amino acids based on functional group with examples (No structures required)	3
	b)	Write a note on glycolipids or phospholipids	3
	c)	Discuss the biochemical role of thiamine or nicotinamide	2
	d)	Explain melting and annealing of DNA	2
	e)	Give example of high energy phosphate bond and explain their role	2
5.	a)	Write a note on folic acid or pantothenic acid	3
	b)	Discuss monosaccharides in detail	3
	c)	Write a note on Vitamin-A or Vitamin-D	3
	d)	Draw the structure of arachidonic acid	2
	e)	Write salient features of lipid digestion	1
		NAVI MUMPAL - INDIA	
6.	a)	Write a short note on Vitamin-K or Vitamin-E	3
	b)	Differentiate between oils and fats	3
	c)	Write a note on Double Helix structure of DNA	3
	d) ·	Explain laws of thermodynamics	2
	e).	Define Iodine value	1



Hode

[10M]Q.3] Confused Pairs of Words: 1. Affect- Effect 2. Complement-Compliment 3. Principal-Principle 4. Sight - Site 5. Accept - Except [10M] Q.4 Define the following: { Any Five } 1. Communication along with its complete cycle 2. Gesture 3. Cell Phone Etiquette 4. Resume 5. Feedback 6. Stress Management [15M] Q.5] Write Short Notes on: { Any Three } 1. Barriers of Communication 2. Describe the Methods Of Communication 3. Types of Letters 4. Importance of Technical Communication and its different types 5. Flow of Communication in Organization 6. Impact of Media & Technology 7. Family Responsibilities 10M 0.6] Write an ESSAY on any one of the following topics: 1. "Pharmacy Day", Celebration in the campus 2. "Field Project" that you did for Communication Skills Subject 3. "Health is Wealth" Discuss 4. Discuss how you will Communicate with Cancer Patient & Mentally Challenged Q.7] Draft a Cover Letter and Resume for the post of an Assistant Pharmacist in Cipla [10M] Company [10M] 0.8] Attempt any One:

(a) You want to go abroad for your Higher studies in Pharmacy. Write an Inquiry Letter to Citibank inquiring about an educational loan.

OR

(b) You are dissatisfied with the garments consignment you have received from a supplier.

Write down a Complaint Letter for the same and mention the details regarding the problem.